

DESIGN, SYNTHESIS, AND CHARACTERIZATION OF A STIMULI-RESPONSIVE PEPTIDE-BASED HYDROGEL



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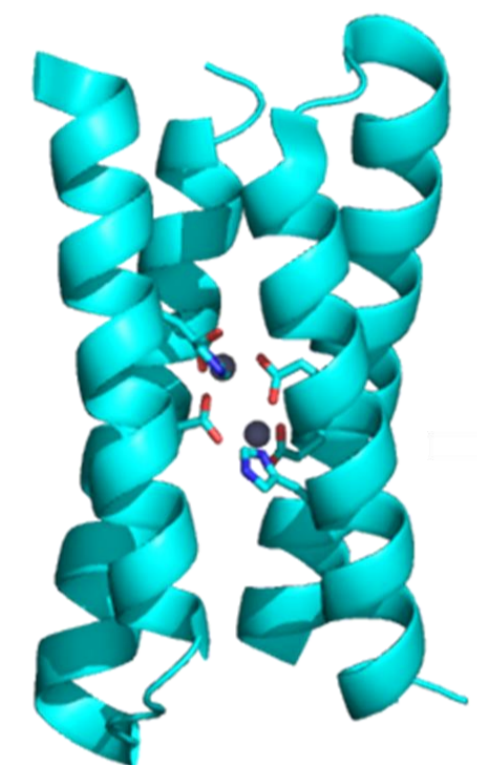
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INTRODUCTION

Extreme weather events are driving the search for biosafe strategies for plant protection. **Peptide-based hydrogels** provide a sustainable solution, combining water retention, controlled bioactive release, and stimulus responsiveness¹⁻³. **Metal coordination** has been widely used to guide peptide and protein **self-assembly** into well-defined nanostructures⁴⁻⁵. Here, we show that peptides inspired by *Due Ferri* metalloproteins (**hDF**, **hDF2**) can form fibrillar hydrogel networks via metal coordination and hydrophobic interactions, tunable by pH, temperature, peptide ratio, and metal ions, with potential biomedical and environmental applications.

Design of hDF



DFI (PDBid: 1ec5)

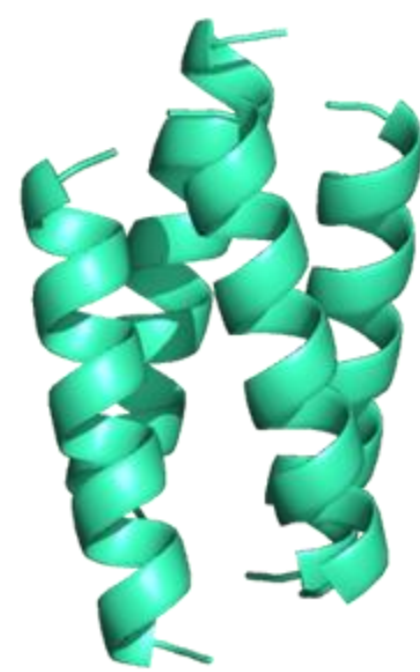
Antiparallel four-helix bundle with two metal binding sites

aGlu-x-x^dHis motif involved in the metal binding

Starting backbone coordinates

Extracted backbone

Application of the Crick equation and frequency fitting
26-residue helix as building unit

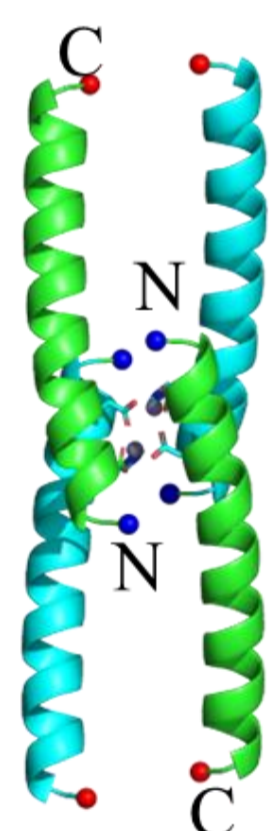
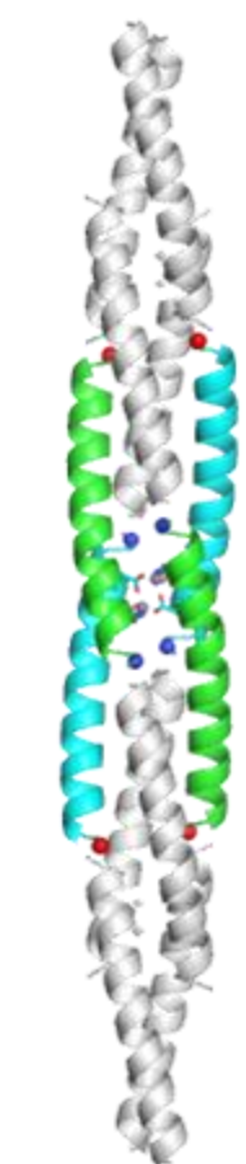


Regenerated backbone

The CCCP 'structure generator' creates a **68-residue tetrameric coiled-coil scaffold**

hDF fiber

Fibrillar poly-Gly cut into **antiparallel staggered heterotetramers** with N-terminal binding site and C-terminal hydrophobic core to allow **self-assembly**



hDF heterotetramer

1st and 2nd coordination shell residues match DFI

'a' and 'd': Leu/Ile to promote an antiparallel topology

'f': Ala to enhance helicity and support hydrogelation

b/e,c/g: +/- Promote desired and prevent undesired topologies

DFI-helix A

fg abcdefg abcdefg abcdefg abcdefg a

hDF-p1

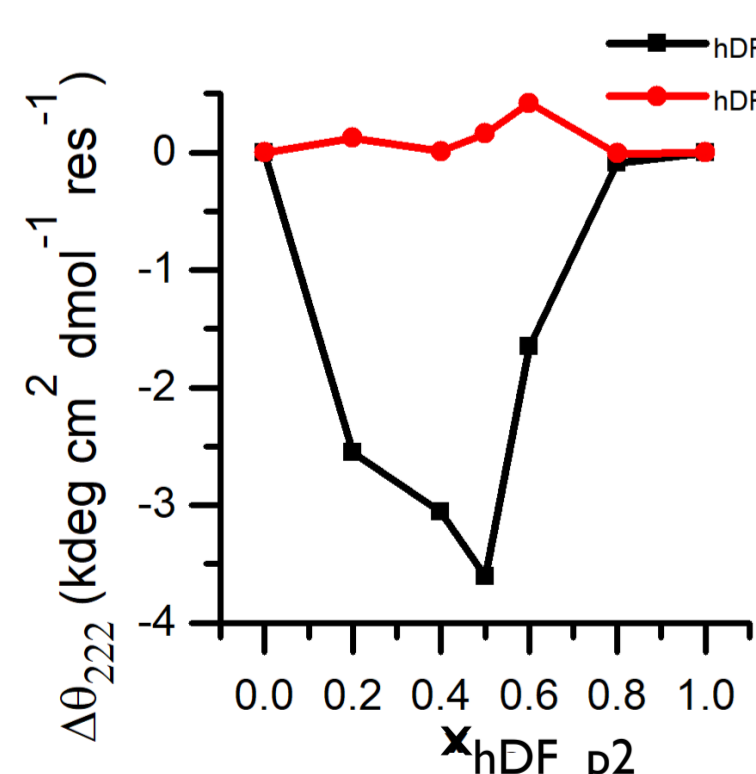
DY LRELLKL ELQLIKQ YREALY V
LEAL EEQLKAE YKSALAW LKAIAKAK L

DFI-helix B

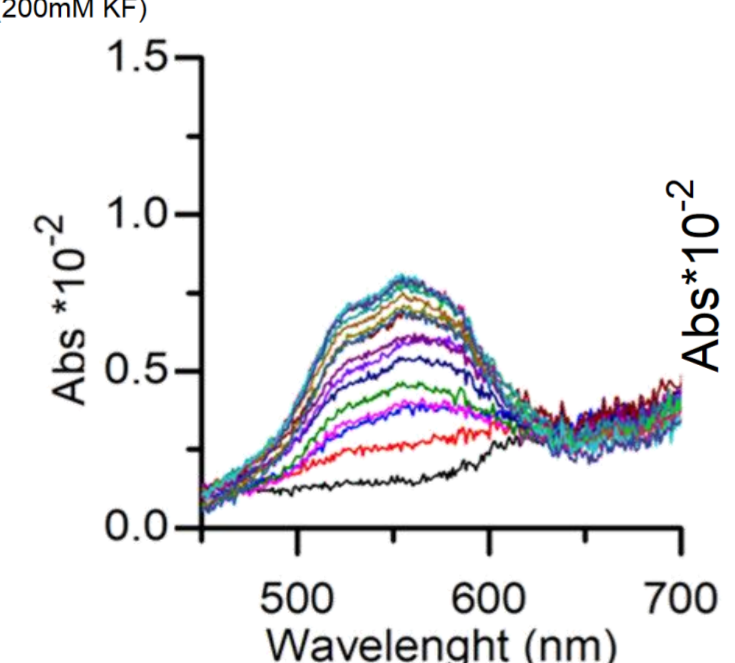
PV LAKILED EEKHIEW LETILG

hDF-p2

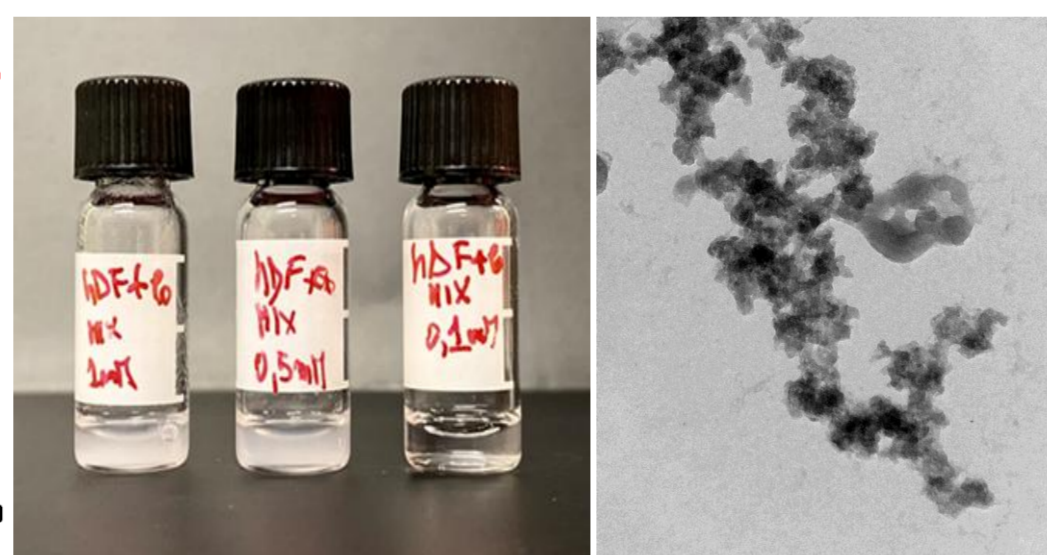
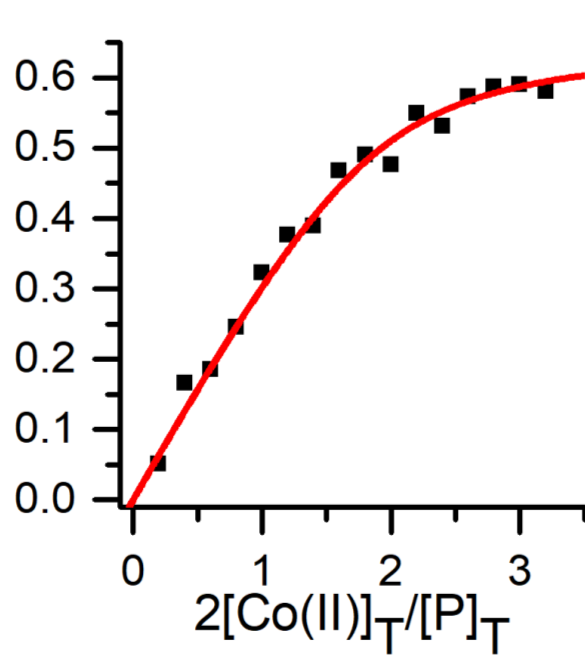
IEAD EEKHAE LKSILAW LEAIEAK L



hDF-p1 and hDF-p2 recognize each other (1:1), **KF** prevents salt bridges



UV-Vis titration abs at 550 nm increased linearly up to 2:1 **Co(II):tetramer**



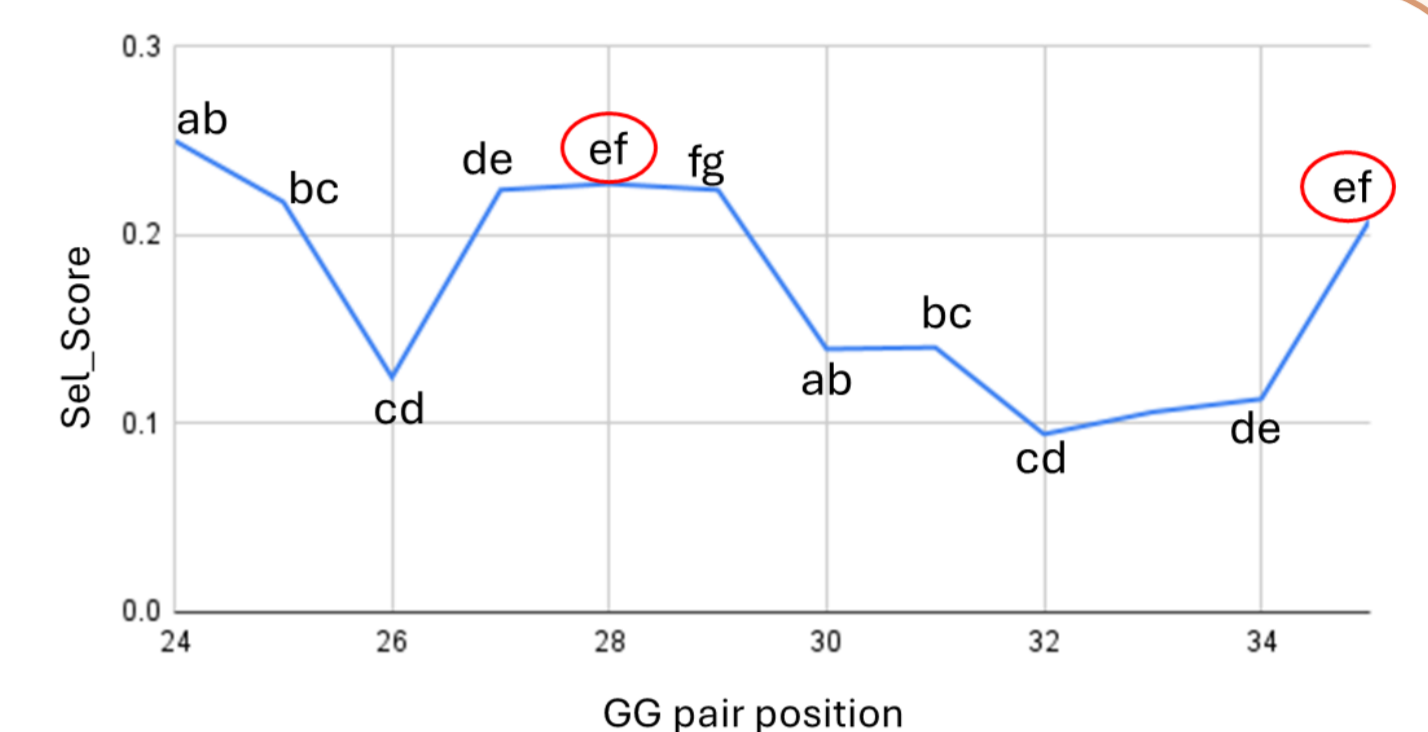
Amorphous precipitate Too fast statistical assembly of fibers

Re-design of hDF2

hDF → hDF2

'c-g' and 'b-e' pairs were mutated

GG in 'e' and 'f' positions mimics the chain break and favors a staggered antiparallel heterotetramer



DFtet-A

g abcdefg abcdefg abcdefg abcdefg abcd

hDF2-p1

K LKELKSK LKELLKL ELQAIKQ YKELKAE LKEL

hDF2-p1.2

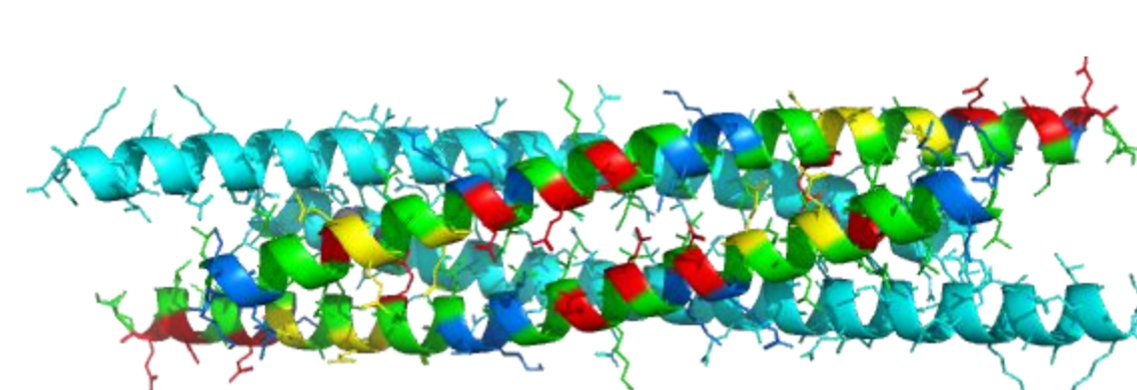
K LKLLAL EQQLAQ YREIKAE LKEIGG

DFtet-B

E LEELESE LEKILED EERHIEW LEKLEAK LKEL

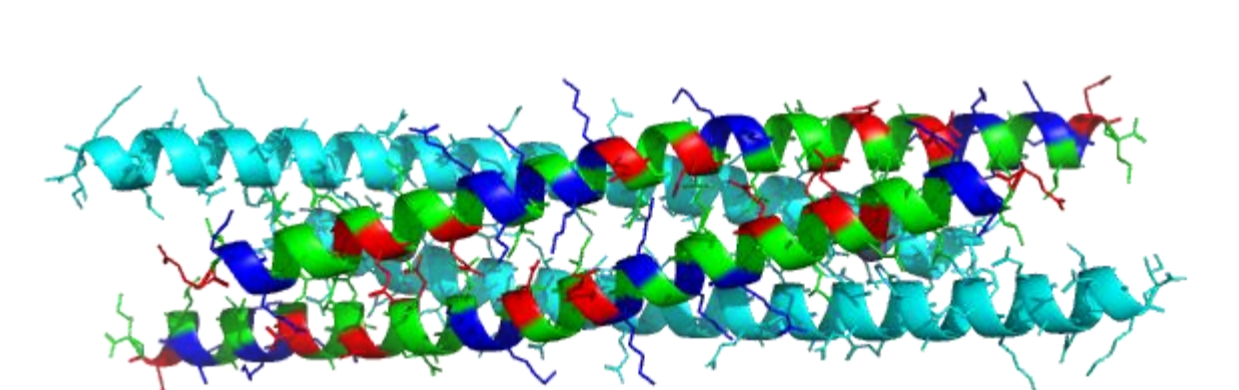
hDF2-p2

K LEEIEAD EERHIAW LQKIEAE LEKIGG



hDF2-p1 + hDF2-p2

improved salt bridge interactions



hDF2-p1.2 + hDF2-p2

Gelation tests

Macroscopic hydrogel formation assessed by **inverted vial tests** with different pH, metals and buffers

SOL-GEL TRANSITIONS

hDF2-p2

1 mM peptide in **MOPS** pH 7.0 + **Zn(II)** on ice



Highly pH-dependent system

hDF2-p1.2 + hDF2-p2

1 mM **mix** in **PB** pH 5.6 (A), pH 7.0 (B) + **Co(II)** on ice

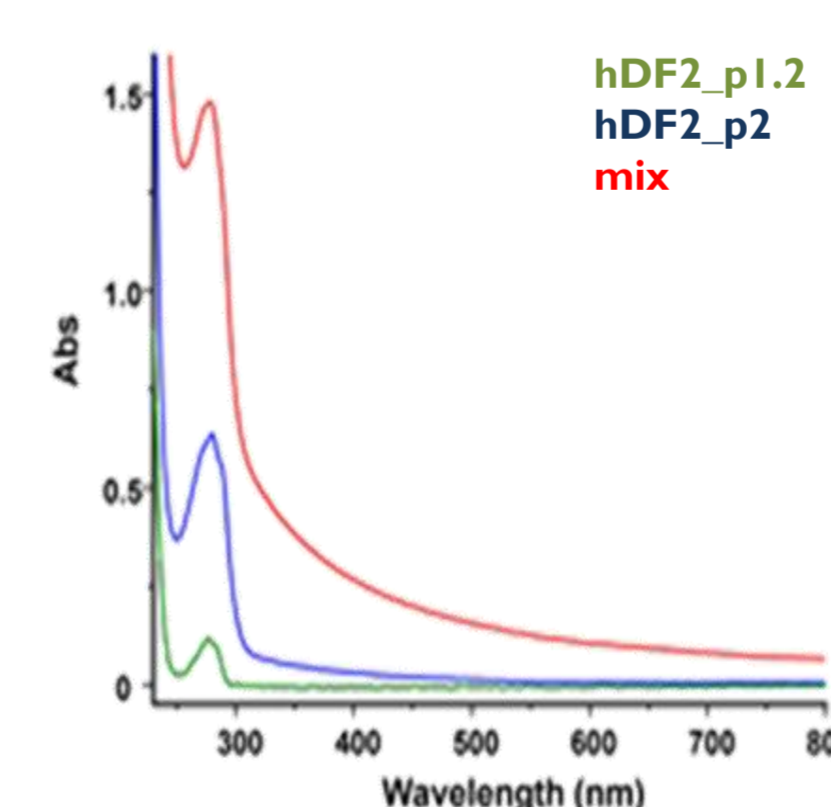


A

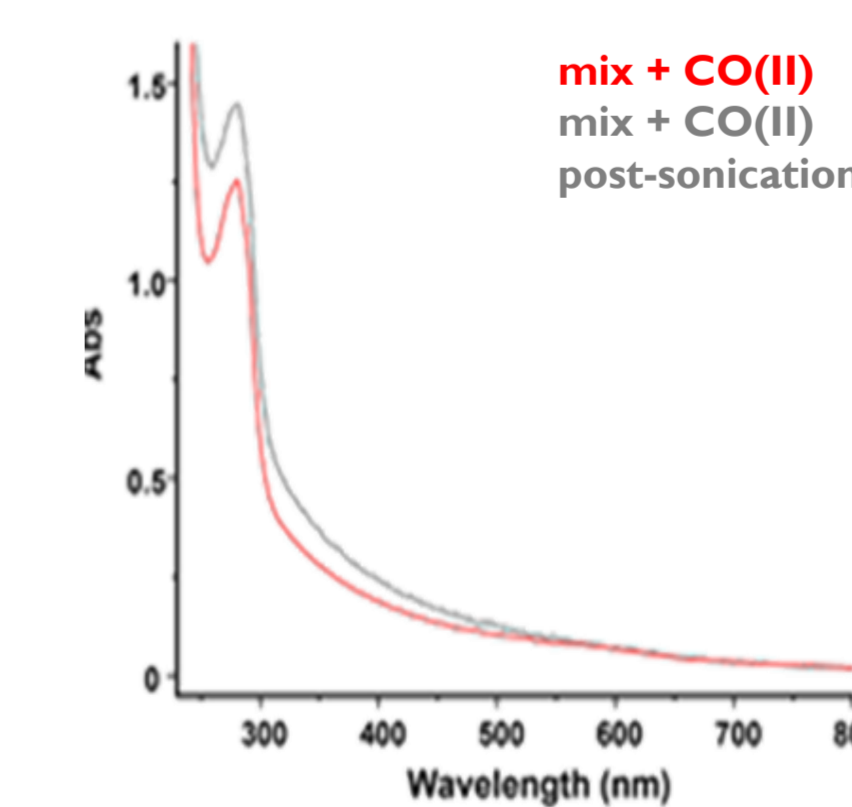
Hydrogel formation is metal-dependent and pH-independent



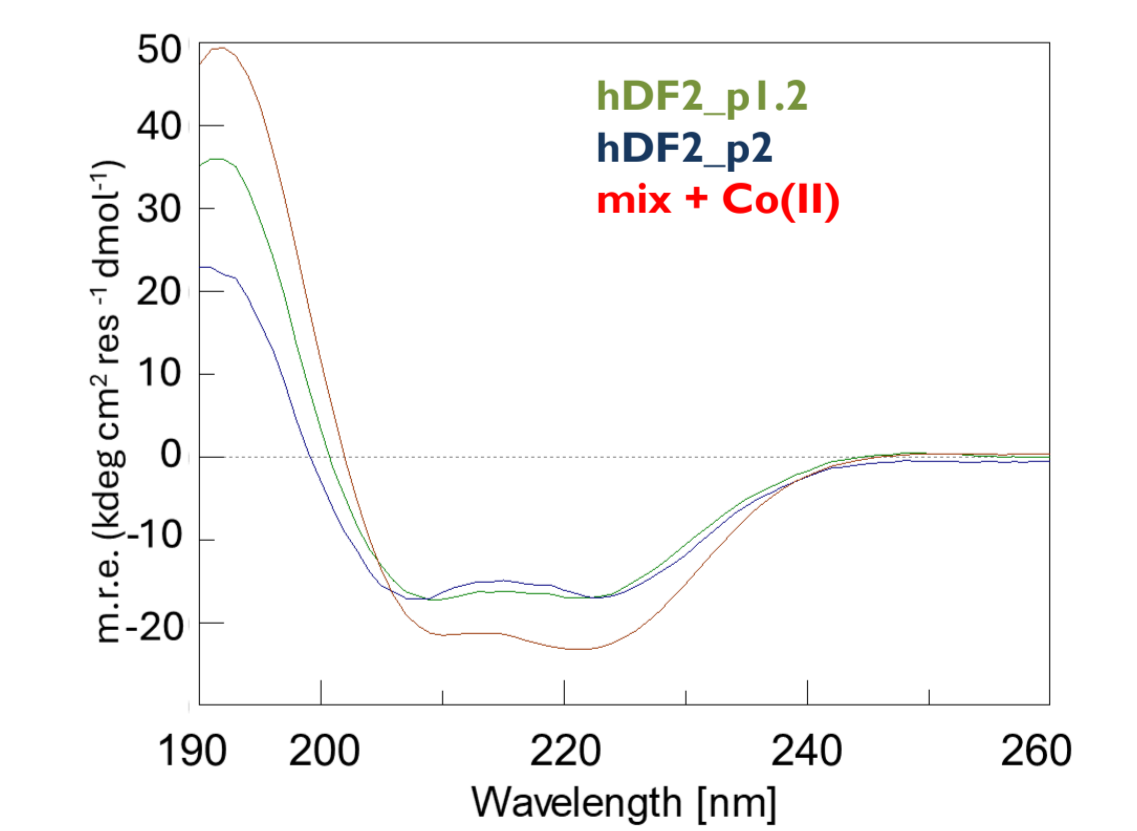
B



Aggregation of hDF2-p1.2 and hDF2-p2 (1:1) Scattering >800 nm



Spectrum unchanged after Co(II) addition



α-helical structures; Red-shift: tighter helices, hydrogel network

CONCLUSIONS AND PERSPECTIVES

Metal-directed self-assembly of peptides inspired by *Due Ferri* metalloproteins enables the formation of tunable hydrogel networks. **hDF2-p2** forms hydrogels under specific pH-dependent conditions, while redesign of **hDF2-p1** enhances its interaction with hDF2-p2, creating a system where hydrogelation is both pH- and metal-dependent. UV-vis and CD analyses confirm peptide aggregation while preserving α-helical structure, consistent with robust hydrogel network formation, highlighting the potential of these materials for environmental applications⁶⁻⁸.

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